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Renewable methanol synthesis from CO₂ with indium-based catalysts formed by pulsed laser ablation in liquids

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PURPOSE OF THE ABSTRACT

With intensifying global warming and depleting fossil fuels, it is desirable to convert anthropogenic CO₂ into valuable chemicals. In this respect, methanol is a promising feedstock for producing commodity chemicals such as olefins and formaldehyde, since its demand has been increasing continuously over the past decade.[1] Moreover, the energy generated from renewable sources has to be stored in certain chemical energy carriers before usage.[2] With a high volumetric energy density (4.33 kWh L⁻¹), methanol is an ideal candidate for chemical energy storage, which can be produced from CO₂ and hydrogen derived from water electrolysis.[3]

Recently, supported In₂O₃-catalysts have shown excellent long time stability and higher methanol selectivity compared to traditional Cu/ZnO-based systems.[4] However, the activity of indium-based catalysts is very susceptible to the preparation method which is typically wet impregnation or co-precipitation.[5] An alternative and innovative catalyst preparation method, namely "pulsed laser ablation in liquids" (PLAL) was successfully applied for several common active metals (Pt, Pd, Au, Cu) in the past.[6] Thereby, the method offered high process control for the design of catalyst parameters like particle size and defined elemental composition.

Here, we explore a laser-based synthesis route to prepare In₂O₃-catalysts of varying indium size and load. To the best of our knowledge, In₂O₃ catalysts prepared by "pulsed laser ablation in liquids" (PLAL) were applied in thermally activated catalysis for the first time. Several batches, each consisting of catalysts with different indium loadings (2-25 wt%) on ZrO₂, were prepared using individual laser synthesis parameters.

By choosing the wavelength of the (infrared or green) laser light, we were able to adjust the size distribution of the In₂O₃-nanoparticles, while the surrounding liquid of the synthesis process influenced their chemical composition. After deposition of In₂O₃ on several commercially available oxidic supports (e.g. TiO₂, CeO₂, ZrO₂) the properties of the catalysts were analyzed using XRD, XPS, H₂ TPR and HR-TEM. Depending on the parameters of the PLAL process, the catalysts showed a distinct difference in crystallinity and morphology.

Subsequently, the catalysts were applied for methanol synthesis from CO₂ in a slurry reactor concept that has shown interesting process features in our previous publication.[7] The choice of the support had a clear effect on the catalytic performance. Although crystallinity and morphology of the indium species were depending on the settings of the laser process, catalytic activity stayed comparably robust. Post-reaction analysis of these catalysts revealed a stepwise convergence of their composition and properties by oxidation and recrystallization processes, explaining the similar activity in CO₂-hydrogenation. Part of these scientific results were published in 2021 in ACS Applied Energy Materials.[8]

FIGURES

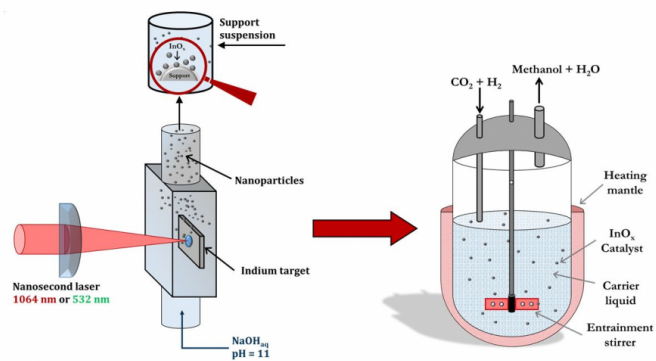


FIGURE 1

Synthesis of InO_x-catalysts by PLAL and their usage for slurry phase CO₂-hydrogenation.

Schematic representation of the PLAL setup and procedure to synthesize supported InO_x nanoparticles.

These particles are applied as catalysts in slurry methanol synthesis afterwards.

FIGURE 2

KEYWORDS

Pulsed laser ablation in liquids | CO₂-hydrogenation | indium catalysts | slurry methanol synthesis

BIBLIOGRAPHY