

N°1255 / PC

TOPIC(s) : Polymers or composites / Alternative solvents

From PEF to rPEF: disclosing the potential of deep eutectic solvents in continuous de-/re-polymerization recycling

AUTHORS

Beatriz AGOSTINHO / CICECO - AVEIRO INSTITUTE OF MATERIALS, DEPARTMENT OF CHEMISTRY, UNIVERSITY OF AVEIRO, CAMPUS DE SANTIAGO, AVEIRO

Andreia F. SOUSA / CICECO - AVEIRO INSTITUTE OF MATERIALS, DEPARTMENT OF CHEMISTRY, UNIVERSITY OF AVEIRO, CAMPUS SANTIAGO, AVEIRO

Armando J. D. SILVESTRE / CICECO - AVEIRO INSTITUTE OF MATERIALS, DEPARTMENT OF CHEMISTRY, UNIVERSITY OF AVEIRO, CAMPUS SANTIAGO, AVEIRO

Corresponding author : Andreia F. SOUSA / andreiafs@ua.pt

PURPOSE OF THE ABSTRACT

With the rise of bio-based polymers' research and production, there is an emerging concern about their fate after usage, since some of them also tend to be persistent in the environment. Poly(ethylene 2,5-furandicarboxylate) (PEF), derived from C6 sugars like D-fructose or D-glucose [1], is widely known to be a viable replacement for conventionally used poly(ethylene terephthalate) (PET), mainly due to its relevant thermal and mechanical properties [2-5] as well as standout barrier features, with a 19 fold reduction in CO₂ permeability compared to PET [6]. Despite its great potential and expected global production [7], its end-of-life (EoL) options are generally overlooked, as we have recently highlighted [8]. In this work, we report the design of urea:zinc acetate (U:ZnAc₂) Deep Eutectic Solvent (DES) system for a continuous, mild and closed-loop recycling approach from PEF into rPEF (Figure 1). First, PEF was depolymerized by mild glycolysis using U:ZnAc₂ (4:1 molar ratio) as a catalyst. Afterwards, the reaction mixture was repolymerized without any purification steps or additional catalysts in a novel one-pot approach, yielding again PEF (rPEF), as confirmed by ¹H and ¹³C NMR, with a 69 % yield. When a co-catalyst was added, titanium(IV) tert-butoxide (TBT), a maximum yield of 91 % was achieved.

For the process' optimization, a series of Lewis Acid ESs were screened and tested for their efficiency in PEF depolymerization, and the reaction conditions were optimized to maximize the initial polymer's weight loss.

Additionally, if one wants to isolate intermediate products, that can be achieved by precipitation with water, allowing to confirm that bis(hydroxyethyl)-2,5-furandicarboxylate (BHEFDC) is the main intermediate product formed. The proposed recycling approach confirms, for the first time, the potential for DESs to catalyze PEF's de-/re-polymerization in a continuous way, as an efficient greener option to chemically recycle persistent PEF wastes (and beyond), promoting a more circular approach for its end-of-life.

FIGURES

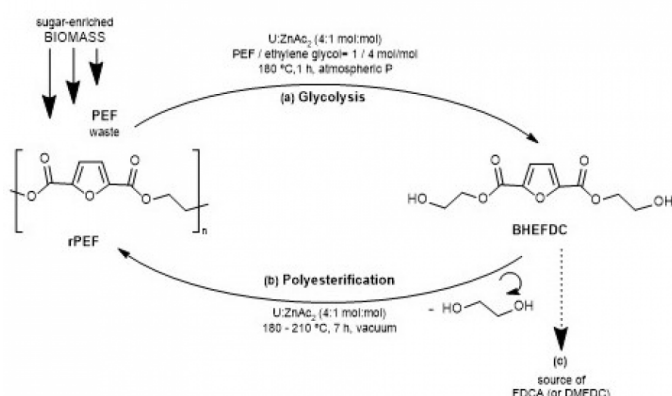


FIGURE 1

PEF continuous circular recycling method

PEF continuous circular recycling method through a DES-assisted mild glycolysis (a) and polyesterification (b) reactions, with potential recovery of FDCA (or its dimethyl ester) (c) also considered.

FIGURE 2

KEYWORDS

Poly(ethylene 2,5-furandicarboxylate) (PEF) | Deep Eutectic Solvent (DES) | Chemical Recycling | Polymer's End-of-Life

BIBLIOGRAPHY

- [1] K. Loos, R. Zhang, I. Pereira, B. Agostinho, H. Hu, D. Maniar, N. Sbirrazzuoli, A. J. D. Silvestre, N. Guigo and A. F. Sousa, *Frontiers in Chemistry*, 2020, 8, 1-18;
- [2] C. F. Araujo, M. M. Nolasco, P. J. Ribeiro-Claro, S. Rudic, A. J. Silvestre, P. D. Vaz and A. F. Sousa, *Macromolecules*, 2018, 51, 3515-3526;
- [3] A. F. Sousa, C. Vilela, A. C. Fonseca, M. Matos, C. S. R. Freire, G. J. M. Gruter, J. F. J. Coelho and A. J. D. Silvestre, *Polymer Chemistry*, 2015, 6, 5961-5983.;
- [4] A. Gandini, A. J. D. Silvestre, C. Pascoal Neto, A. F. Sousa and M. Gomes, *Journal of Polymer Science Polymer Chemistry*, 2009, 5, 295-298;
- [5] R. J. I. Knoop, W. Vogelzang, J. Van Haveren and D. S. Van Es, *Journal of Polymer Science Part A: Polymer Chemistry*, 2013, 51, 4191-4199;
- [6] S. K. Burgess, R. M. Kriegel and W. J. Koros, *Macromolecules*, 2015, 48, 2184-2193;
- [7] Avantium, YXY Technology, <https://www.avantium.com/technologies/yxy/>, (accessed 20 July 2021);
- [8] A. F. Sousa, R. Patrício, Z. Terzopoulou, D. N. Bikiaris, T. Stern, J. Wenger, K. Loos, N. Lotti, V. Siracusa, A. Szymczyk, S. Paszkiewicz, K. S. Triantafyllidis, A. Zamboulis, M. S. Nikolic, P. Spasojevic, S. Thiagarajan, D. S. Van Es and N. Guigo, *Green Chemistry*, 2021, 23, 8795-8820.