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Extraction and selective recovery of transition metals from acid mine drainage

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PURPOSE OF THE ABSTRACT

The fast development of the electronic industry has led to an ever-increasing global demand in metals and has played an important role in nowadays metal scarcity. The development of processes that allow not only the environment detoxification but also the recycling of waste streams for the recovery of valuable metals, in line with the principles of a circular economy, becomes imperative. In this work, ionic-liquid-based aqueous biphasic systems (IL-based ABS) were used as a more benign alternative to conventional liquid-liquid extraction processes in transition metals (Zn, Cu, Co and Ni) selective and sequential recovery from acid mine drainage (AMD) waters.[1] Due to their ionic nature, the use of ILs allows metal extraction mechanisms that are not possible when conventional organic solvents are applied.[2] Thus, ABS composed of different ILs and Na₂SO₄ were evaluated and it was observed that ILs anion presents an important role in metal extraction, with thiocyanate anion (SCN) presenting a remarkable ability to extract Zn, Cu and Co from AMD through the formation of stable metal complexes. It was also shown that the use of NaSCN salt as an additive in IL-based ABS allows to reduce system cost and environmental impact but also to tune metal selectivity by the concentration of SCN anion added to the ABS. Furthermore, the formation of a hydrophobic salt composed of IL cations and metal-SCN complexes anions allowed the selective extraction and recovery of copper and zinc in a single step. These results permitted the development of a sustainable process for the selective and sequential recovery of 73.1 % of zinc (98.8 % pure) and 46.3 % of copper (87 % pure) present in pre-treated AMD, with no need of extra recovery steps and considering liquid phases recovery and recycling.

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FIGURES

FIGURE 1

FIGURE 2

KEYWORDS

aqueous biphasic systems | recycling | selective separation | circular economy

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