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TOPIC(s) : Alternative solvents

Using an in vivo model to assess NADES toxicity

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PURPOSE OF THE ABSTRACT

Deep Eutectic Systems (DES) are part of a new class of green solvents¹ and have been coined as biocompatible and biodegradable². DES can be defined as a combination of two or more components where one is a hydrogen bond acceptor (HBA) and the other a hydrogen bond donor (HBD). One relevant physical feature of these systems is the fact that the melting point of the mixture is lower than the melting point of any of the components individually^{3,4}. Natural Deep Eutectic Systems (NADES) are a subclass of DES, where the compounds are all naturally occurring molecules⁵, such as betaine, citric acid, glucose and lactic acid. Nevertheless, currently, the number of studies involving toxicity assessment is very limited and in addition, each system is unique and needs to be taken into consideration. The main goal of this work was to assess the toxicity of three different NADES (Citric acid: Trehalose: Water (2:1:3), Betaine: Glycerol (1:2) and Betaine: Sorbitol: Water (1:1:3)) by injecting them intraperitoneally in zebrafish (*Danio rerio*) at four different concentrations. This is a biological model often used in pharmacological and environmental trials. In this work, the activity of different enzymes involved in antioxidant pathways or xenobiotic elimination and lipid peroxidation were assessed. The results suggest that NADES do not show significant toxicity at the tested concentrations. The promising results obtained here suggest that these systems have the potential to be employed as a new class of green solvents that can be used in several applications, for example, pharmaceutical and cosmetic industry and cell cultures without harming living organisms.

FIGURES

FIGURE 1

FIGURE 2

KEYWORDS

DES | toxicity | zebrafish

BIBLIOGRAPHY

- [1] Gotor-Fernández V, Paul CE. Deep eutectic solvents for redox biocatalysis. Vol. 293, Journal of Biotechnology. Elsevier B.V.; 2019. p. 24-35.
- [2] Abbott AP, Capper G, Davies DL, Rasheed RK, Tambyrajah V. Novel solvent properties of choline chloride/urea mixtures. Chemical Communications [Internet]. 2003 Jan 1 [cited 2021 Jun 2];(1):70-1. Available from: <http://www.rsc.org/suppdata/cc/b210714g/>
- [3] Abbott AP, Boothby D, Capper G, Davies DL, Rasheed RK. Deep Eutectic Solvents formed between choline chloride and carboxylic acids: Versatile alternatives to ionic liquids. Journal of the American Chemical Society [Internet]. 2004 Jul 28 [cited 2021 Nov 11];126(29):9142-7. Available from: <https://pubs.acs.org/doi/full/10.1021/ja048266j>
- [4] Paiva A, Craveiro R, Aroso I, Martins M, Reis RL, Duarte ARC. Natural deep eutectic solvents - Solvents for the 21st century. ACS Sustainable Chemistry and Engineering. 2014 May 5;2(5):1063-71.
- [5] Choi YH, van Spronsen J, Dai Y, Verberne M, Hollmann F, Arends IWCE, et al. Are natural deep eutectic solvents the missing link in understanding cellular metabolism and physiology? Plant Physiology [Internet]. 2011 Aug [cited 2021 Apr 22];156(4):1701-5. Available from: [/pmc/articles/PMC3149944/](https://pubmed.ncbi.nlm.nih.gov/2149944/)